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(FILE 'HOME' ENTERED AT 07:55:07 ON 16 SEP 2002)

FILE 'CA' ENTERED AT 07:55:23 ON 16 SEP 2002

E SMITH KIM R/AU
 L1 86 S E3-E4
 E FUK PONG MAN V/AU
 E MAN VICTOR FUK PONG/AU
 L2 26 S E2-E3
 E WISETH WENDY/AU
 L3 5 S E3-E4
 E STARDIG RICHARD/AU
 L4 24188 S (HYDRAT? OR MONOHYDRATE# OR DIHYDRATE# OR TRIHYDRATE# OR
 TETR
 L5 313891 S HYDRATABLE OR ANIONIC OR NONIONIC OR ETHOXY? OR EO OR
 DODECYL
 L6 853 S L4 AND L5
 L7 78046 S (WITHOUT OR ABSENT OR ABSENCE OR FREE OR NO OR
 INTERNAL) (5A) (
 L8 5 S L6 AND L7
 L9 176 S L4 AND L7
 L10 13 S L9 AND (DETERGENT# OR WASH? OR WAREWASH? OR DISH? OR
 TABLEWAR
 L11 4105 S (HEXAHYDRATE# OR HEPTAHYDRATE# OR OCTAHYDRATE# OR
 DECAHYDRATE
 L12 1 S L11 AND L5 AND L7
 L13 26 S L11 AND L7
 L14 4 S L13 AND (DETERGENT# OR WASH? OR WAREWASH? OR DISH? OR
 TABLEWA
 L15 32 S (H2O) (4A) (NA2CO3 OR NA2SO4 OR STPP OR NA2HCO3 OR SILICATE#
 OR
 L16 2 S L15 AND L7
 L17 5 S HYDRAT? (P) HYDRATABLE (P) (MOLD? OR MOULD?) AND (DETERGENT# OR
 L
 L18 30 S (MOULD? OR MOLD?) (P) (ENZYME# OR PROTEASE# OR AMYLASE# OR
 LIPA
 L19 1 S L18 AND L4
 L20 0 S L18 AND L11
 L21 14 S (MOULD? OR MOLD?) (P) (GLYCOL ETHER# OR BUTOXYETHANOL OR
 BUTOXY
 L22 395064 S EXTRUD? OR EXTRUSION OR CAST?
 L23 359 S L4 AND L22
 L24 7 S L23 AND L7
 L25 46 S L11 AND L22
 L26 293 S (EXTRUD? OR EXTRUSION) (P) (ENZYME# OR PROTEASE# OR AMYLASE#
 OR
 L27 3 S L4 AND L26
 L28 0 S L11 AND L26
 L29 0 S L15 AND L26
 L30 6 S (EXTRUD? OR EXTRUSION) (P) (ENZYME# OR PROTEASE# OR AMYLASE#
 OR

FILE 'STNGUIDE' ENTERED AT 09:13:42 ON 16 SEP 2002

FILE 'CA' ENTERED AT 09:23:44 ON 16 SEP 2002

FILE 'USPATFULL' ENTERED AT 09:24:04 ON 16 SEP 2002

L31	1219 S L10
L32	416 S L31 AND (MOLD? OR MOULD?)
L33	80 S L13 AND DETERGENT#
L34	20 S L33 AND L22
L35	41 S L18
L36	58 S L27
L37	11 S L30

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L8 5 L6 AND L7

=> d 1-5 l8 ti

L8 ANSWER 1 OF 5 CA COPYRIGHT 2002 ACS

TI Organic-mineral modifier for cementitious systems

L8 ANSWER 2 OF 5 CA COPYRIGHT 2002 ACS

TI Liquid nonaqueous cleaning products containing a bleach and a free
radical
 scavenger

L8 ANSWER 3 OF 5 CA COPYRIGHT 2002 ACS

TI Preparation and properties of copper synthetic **anionic** clays

L8 ANSWER 4 OF 5 CA COPYRIGHT 2002 ACS

TI Anodic electrocoating compositions

L8 ANSWER 5 OF 5 CA COPYRIGHT 2002 ACS

TI Possibility of removing water of crystallization from pentaammonium
tripolyphosphate monohydrate without destroying the
anion

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> d 1-13 l10 ti

L10 ANSWER 1 OF 13 CA COPYRIGHT 2002 ACS

TI Preparation of sodium **perborate hydrate** granules
without heating

L10 ANSWER 2 OF 13 CA COPYRIGHT 2002 ACS

TI Liquid nonaqueous cleaning products containing a bleach and a free
radical
scavenger

L10 ANSWER 3 OF 13 CA COPYRIGHT 2002 ACS

TI Manufacture of sodium **carbonate monohydrate**

L10 ANSWER 4 OF 13 CA COPYRIGHT 2002 ACS

TI Calcium silicate in the form of .alpha.-Wollastonite and its
intermediates

L10 ANSWER 5 OF 13 CA COPYRIGHT 2002 ACS

TI Bleaching and cleaning mixtures

L10 ANSWER 6 OF 13 CA COPYRIGHT 2002 ACS

TI Dipierisophthalic acid bleaching compositions

L10 ANSWER 7 OF 13 CA COPYRIGHT 2002 ACS

TI Calcium **sulfate dihydrate** from mineral
phosphate

L10 ANSWER 8 OF 13 CA COPYRIGHT 2002 ACS

TI High-purity cerous sulfate octahydrate

L10 ANSWER 9 OF 13 CA COPYRIGHT 2002 ACS

TI Synthetic zeolite

L10 ANSWER 10 OF 13 CA COPYRIGHT 2002 ACS

TI Utilization of calcined magnesite in neutralization filters

L10 ANSWER 11 OF 13 CA COPYRIGHT 2002 ACS

TI Separation of alkylation catalyst from the products by dialysis

L10 ANSWER 12 OF 13 CA COPYRIGHT 2002 ACS

TI The determination of silicic acid in portland cement and in soluble
silicates by the rapid method

L10 ANSWER 13 OF 13 CA COPYRIGHT 2002 ACS

TI Beryllium. I. Beryllium **sulfate** and its **hydrates**

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L13 ANSWER 7 OF 26 CA COPYRIGHT 2002 ACS

AN 127:163479 CA

TI Detergent compositions that solidify **without heat**,
pressure or water

IN Ando, Yoshitaka; Hiki, Kiyotaka

PA Teii Hooru K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C11D007-06

ICS C09K003-00; C11D007-14; C11D007-16; C11D017-06

CC 46-6 (Surface Active Agents and Detergents)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	JP 09176691	A2	19970708	JP 1995-333633	19951221
	JP 3302549	B2	20020715		
AB	The title compns. contain metal ion sequestering agents and detergents,				
in	the form of solid particles, and at least a portion of the detergent is				
	used in hydrated form. A compn. (100 g) from Na tripolyphosphate 30, Na				
	metasilicate-9H ₂ O 10, and NaOH 42 parts was allowed to solidify in a				
	500-mL plastic container for 1 day to give a detergent showing good				
	detergency in dishwashing.				
ST	dishwashing detergent sequestering agent				
IT	Detergents				
	Sequestering agents				
	(detergent compns. that solidify without heat ,				
	pressure or water)				
IT	Detergents				
	(dishwashing; detergent compns. that solidify without				
	heat , pressure or water)				
IT	60-00-4, EDTA, uses 497-19-8, Sodium carbonate, uses 1310-73-2,				
Sodium	hydroxide, uses 1344-09-8, Sodium silicate 5064-31-3, Trisodium				
	nitrilotriacetate 6132-02-1, Sodium carbonate				
	decahydrate 7727-73-3, Sodium sulfate				
	decahydrate 7757-82-6, Sodium sulfate, uses 7758-29-4, Sodium				
	tripolyphosphate 13517-24-3, Sodium metasilicate nonahydrate				
RL:	TEM (Technical or engineered material use); USES (Uses)				
	(detergent compns. that solidify without heat ,				
	pressure or water)				

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L13 ANSWER 6 OF 26 CA COPYRIGHT 2002 ACS
 AN 127:207327 CA
 TI Solid detergent and its manufacturing method
 IN Ando, Yoshitaka; Hiki, Kiyotaka
 PA T. Paul K. K., Japan
 SO Jpn. Kokai Tokkyo Koho, 12 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 IC ICM C11D017-00
 ICS C11D007-14; C11D007-16
 CC 46-6 (Surface Active Agents and Detergents)
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 09217100	A2	19970819	JP 1996-25207	19960213

AB The detergent esp. useful for dishwashing is manufd. by prepg. a uniformly mixed detergent compn. that partially contains hydrate compds. from an aggregate of solid particles with av. particle size 0.05-2.0 mm and naturally solidifying the mixt. under **no heating** and **no** pressuring. Prepg. a compn. (av. particle size 0.8 mm) contg. Na tripolyphosphate 30, Na metasilicate-9 H2O 10, and NaOH 42%, filling the compn. in a closed container, and sitting at 20-25.degree. for 1 day gave a solid detergent with sp. gr. 1.4, good detergency and solidification degree.

ST solid detergent solidification dishwashing; hydrate detergent solidification

IT Detergents
 (dishwashing; solid detergent prepn. by solidification with hydrate compds.)

IT Solidification
 (solid detergent prepn. by solidification with hydrate compds.)

IT Hydrates
 RL: MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (solid detergent prepn. by solidification with hydrate compds.)

IT 6132-02-1, Sodium **carbonate decahydrate** 7727-73-3, Sodium **sulfate decahydrate** 13517-24-3, Sodium silicate nonahydrate
 RL: MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (solid detergent prepn. by solidification with hydrate compds.)

IT 7758-29-4, Sodium tripolyphosphate
 RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 (solid detergent prepn. by solidification with hydrate compds.)

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L14 4 L13 AND (DETERGENT# OR WASH? OR WAREWASH? OR DISH? OR
TABLEWARE#

)

=> d 1-4 l14 ti

L14 ANSWER 1 OF 4 CA COPYRIGHT 2002 ACS

TI Solid **detergent** and its manufacturing method

L14 ANSWER 2 OF 4 CA COPYRIGHT 2002 ACS

TI **Detergent** compositions that solidify **without**
heat, pressure or water

L14 ANSWER 3 OF 4 CA COPYRIGHT 2002 ACS

TI Preparation of zinc sulfate and active zinc oxide from zinc ore by wet
process

L14 ANSWER 4 OF 4 CA COPYRIGHT 2002 ACS

TI High-purity cerous **sulfate octahydrate**

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=> d 1-2 116 ti

L16 ANSWER 1 OF 2 CA COPYRIGHT 2002 ACS

TI Structures of some supposed 2,4-azetidinediones. I. Derivatives of
malonic
acid

L16 ANSWER 2 OF 2 CA COPYRIGHT 2002 ACS

TI The extraction of phenols with sodium sulfide solution

=>

17 5 HYDRAT?(P)HYDRATABLE(P) (MOLD? OR MOULD?) AND (DETERGENT# OR LAUNDRY OR DISH? OR TABLEWARE# OR WASH?)

=> d 1-5 117 ti

L17 ANSWER 1 OF 5 CA COPYRIGHT 2002 ACS

TI Preparation of a solid cleaning and/or disinfecting composition by injecting water into a powder

L17 ANSWER 2 OF 5 CA COPYRIGHT 2002 ACS

TI Method of making solid cast alkaline **detergent** composition

L17 ANSWER 3 OF 5 CA COPYRIGHT 2002 ACS

TI Cast **washing** agent

L17 ANSWER 4 OF 5 CA COPYRIGHT 2002 ACS

TI Cast **detergent**-containing article.

L17 ANSWER 5 OF 5 CA COPYRIGHT 2002 ACS

TI Concrete-like products from wastes of the soda industry

=> d 1-5 117 hit

L17 ANSWER 1 OF 5 CA COPYRIGHT 2002 ACS

AB A powd. cleaning and/or disinfecting compn. contg. .gtoreq. 1 water-sol. or **hydratable** component is placed in a **mold**, and water or an aq. soln. or dispersion of additives is injected at 0-30.degree. into the powder through needles to cause the powder to form a solid **molding**. The process avoids degrdn. of heat-sensitive components of the compn. A mixt. of NaOH particles 40, powd. Na₅P₃O₁₀ 40, powd. Na₂SiO₃.5H₂O 19.5, and Na (Cl₂-14-alkyl)benzenesulfonate 0.5% was placed in a **mold**, and water was injected into the mixt. through needles to give a solid **detergent** compn.

IT **Detergents**

(cleaning compns., molding, manuf. of, by injecting water into powder)

L17 ANSWER 2 OF 5 CA COPYRIGHT 2002 ACS

TI Method of making solid cast alkaline **detergent** composition

AB Particles of a **detergent** component are placed in a **mold**, an aq. soln. of a **hydratable** alk. compd. is added to fill the spaces between the particles, and the soln. is solidified to give a **detergent** block which is useful in a dispenser which functions by spraying water on the block to form a concd. **detergent** soln. for addn. to a **wash** tank, e.g., in a **dishwashing** app. Placing 2 kg particles comprising 50% tripolyphosphate (20-40 mesh) and 50% Na disilicate (20-100 mesh) in a 3.5-L receptacle and adding 2 kg 70% aq. NaOH (at 180.degree.F) gave a solid block after 15 min, 62.3% of the tripolyphosphate being present in the block in unreverted form.

ST tripolyphosphate cast **detergent** block; silicate cast

detergent block; disilicate cast **detergent** block;

dishwashing cast **detergent** block; alkali cast

detergent block; casting **detergent** block alkali

IT **Detergents**

(alk., prepn. of cast blocks of, for **dishwasher** dispensers)

IT 1310-73-2, Sodium hydroxide, uses and miscellaneous 7758-29-4, Pentasodium tripolyphosphate 13870-28-5, Sodium disilicate

RL: USES (Uses)

(**detergent** blocks contg., prepn. of cast, for
dishwasher dispensers)

L17 ANSWER 3 OF 5 CA COPYRIGHT 2002 ACS

TI Cast **washing** agent

AB A three-dimensional, solid, cast, **hydrated** alk.

detergent compn. is prepd. which contains .gtoreq.30%

hydratable alkali metal hydroxide, a sequestering agent for

hardness ions, and water of **hydration**. In some cases, the

compns. contain antifoaming agents, a source of available Cl, and/or

other

additives. The compns. are useful in **dishwashing** machines.

Thus, a 57% aq. NaOH soln. was prepd. at 55-60.degree., mixed with

Na5P3O10, cast into **molds**, and solidified to give a product

contg. NaOH 36.5, water 27.5, and Na5P3O10 36.0%.

ST sodium hydroxide **detergent dishwashing**;

dishwashing detergent alkali solid; tripolyphosphate

dishwashing detergent; hydrate alkali **detergent**

dishwashing

IT Antifoaming agents

Bleaching agents

(**detergents** contg., solid, hydrated, for mech.

dishwashers)

IT **Detergents**

(**dishwashing**, alkali-contg., solid, cast, hydrated)

IT 1310-73-2, uses and miscellaneous 2893-78-9 6834-92-0 7758-29-4
13840-33-0

RL: TEM (Technical or engineered material use); USES (Uses)

(**detergents** contg., solid, hydrated, for mech.

dishwashers)

L17 ANSWER 4 OF 5 CA COPYRIGHT 2002 ACS

TI Cast **detergent**-containing article

AB Solid cast **detergent** compns. in disposable **molds** are

prepd. for use in automatic **dishwashers**, etc. The

detergent compn. is dispensed from the **mold** by a liq.

spray. The cast **detergent** compns. contain water and .gtoreq.1

hydratable chem. In some cases, the cast **detergent**

compns. contain .gtoreq.1 preformed plug or core contg. a Cl source, a

defoamer, etc.. The cast **detergent** compns. minimize Cl

instability and differential soly. problems. Thus, 55 parts 50% aq. NaOH

was heated to 55-60.degree., mixed with nine parts NaOH and 36 parts

Na5P3O10, and cast into a disposable **mold** to prep. a solid

detergent.

ST molding **detergent** automatic dispensing; **dishwasher**

detergent compn molding; bleaching **detergent** compn

molding; cleaning bleaching compn molding; alkali **detergent**

compn molding

IT **Detergents**

(alk., molded, for dispensing by spray liq.)

IT Bleaching agents

(chlorine compds., molded **detergents** contg., for dispensing
by spray liq.)

L17 ANSWER 5 OF 5 CA COPYRIGHT 2002 ACS

AB All the waste products from soda ash manuf., such as overburned and
underburned lenticular lime (A) (contg. CaCO3, CaO, di- and tricalcium
silicate, and clay) and the solids (B) obtained from the distn. residues

are utilized in combination with powd. quartz and, optionally, portland cement, basic blast-furnace slags, H₂O, and a swelling agent to produce bonded concrete-like products by steam-curing. Vertical **molds** of sufficient height, e.g. of >1 story of a building, are used. The complete mixt. contains 15-30% A (contg. CaCO₃ 45-60, CaO 8-20, di- and tricalcium compds. 4-15, and clay 4-8%), 30-50% B (contg. CaCO₃ 50-60,

CaO

6-16, and CaCl₂ <0.45%), and 20-60% quartz (contg. >75% SiO₂), 30-50%

H₂O,

and, if necessary, 5% of a **hydratable** siliceous material and 0.035% of a swelling agent. For example, 30% B in which the CaCl₂

content

had been reduced by **washing**, 30% crushed A sieved from the paste of slaked lime, and 35% quartz powder were fed into a ball mill with the addn. of 5% cement and ground with H₂O to obtain a total moisture content of 45% and a 4900-mesh/sq. cm. residue of 5-10%. The mixt. was agitated, a swelling agent 0.045% Al paste was added, and the mass was poured into vertical **molds** in an autoclave. The mass was steam-cured 2.5

hrs. at 0-8.5, 10.75 hrs. at 8.5, and 0.75 hr. at 8.5-0 atm. The resulting porous concrete had a min. compressive strength of 60-99

kg./sq.

cm. and a tensile-flexural strength of 28 kg./sq. cm., an apparent d. of 0.7-0.8, a free CaO content of 1.7%, and was smooth and free of fissures. After 4 months in H₂O there was no efflorescence.

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=> d 1-14 l21 ti

L21 ANSWER 1 OF 14 CA COPYRIGHT 2002 ACS

TI Polyurethane compositions for molding of transmission belts

L21 ANSWER 2 OF 14 CA COPYRIGHT 2002 ACS

TI Hardener compositions for polyurethanes

L21 ANSWER 3 OF 14 CA COPYRIGHT 2002 ACS

TI Peroxide in binders for foundry molds

L21 ANSWER 4 OF 14 CA COPYRIGHT 2002 ACS

TI Hydrogen peroxide in binders

L21 ANSWER 5 OF 14 CA COPYRIGHT 2002 ACS

TI Urethane resins

L21 ANSWER 6 OF 14 CA COPYRIGHT 2002 ACS

TI Surface treatment of polycarbonate moldings

L21 ANSWER 7 OF 14 CA COPYRIGHT 2002 ACS

TI Heat-insulating articles

L21 ANSWER 8 OF 14 CA COPYRIGHT 2002 ACS

TI Fiber-reinforced reaction resin-plastic foam supporting core-composite materials and moldings

L21 ANSWER 9 OF 14 CA COPYRIGHT 2002 ACS

TI In-the-mold coating

L21 ANSWER 10 OF 14 CA COPYRIGHT 2002 ACS

TI Molded foam materials based on polyisocyanate

L21 ANSWER 11 OF 14 CA COPYRIGHT 2002 ACS

TI Mixture for heat insulation materials

L21 ANSWER 12 OF 14 CA COPYRIGHT 2002 ACS

TI Urethane foams

L21 ANSWER 13 OF 14 CA COPYRIGHT 2002 ACS

TI Internally plasticized epoxy resins

L21 ANSWER 14 OF 14 CA COPYRIGHT 2002 ACS

TI Elastomeric polyurethans

=>

> d 1-7 124 ti

L24 ANSWER 1 OF 7 CA COPYRIGHT 2002 ACS

TI Protective powder coatings having increased **heat-resistance**
which are defect-**free** at greater thickness, preparation thereof
and articles coated therewith

L24 ANSWER 2 OF 7 CA COPYRIGHT 2002 ACS

TI Manufacture of heat-storage capsules

L24 ANSWER 3 OF 7 CA COPYRIGHT 2002 ACS

TI Silica fume-containing, hard setting **castable** refractory
compositions

L24 ANSWER 4 OF 7 CA COPYRIGHT 2002 ACS

TI **Extrusion** molding of calcium **silicate hydrate**
-based materials for lightweight building materials

L24 ANSWER 5 OF 7 CA COPYRIGHT 2002 ACS

TI Thermal insulator for avoiding pinholes in the **casting** of metals

L24 ANSWER 6 OF 7 CA COPYRIGHT 2002 ACS

TI Copper plating on zinc die **castings**

L24 ANSWER 7 OF 7 CA COPYRIGHT 2002 ACS

TI Basic refractory bonding cements resisting hydration

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L30 6 (EXTRUD? OR EXTRUSION) (P) (ENZYME# OR PROTEASE# OR AMYLASE# OR
LIPASE# OR CELLULASE# OR GLYCOL ETHER# OR BUTOXY? OR
GLYCOL#) (P)
(ANIONIC OR NONIONIC OR ETHOXY? OR ALKYL BENZENE OR ABS OR HLAS
OR AES OR SULFONATE# OR SULPHONATE#) (P) (HYDRATE# OR DIHYDRATE#
OR TRIHYDRATE# OR ?HYDRATE)

=> d 1-6 l30 ti

L30 ANSWER 1 OF 6 CA COPYRIGHT 2002 ACS

TI Perfume beads in detergent forms, especially tablets for machine
laundering

L30 ANSWER 2 OF 6 CA COPYRIGHT 2002 ACS

TI Heat-stable antistatic acrylic polymer composition

L30 ANSWER 3 OF 6 CA COPYRIGHT 2002 ACS

TI Extrusion and cutting of detergent compositions to form granules

L30 ANSWER 4 OF 6 CA COPYRIGHT 2002 ACS

TI Polyester and polyamide compositions containing sulfonated polyethers

L30 ANSWER 5 OF 6 CA COPYRIGHT 2002 ACS

TI Curable chlorinated polyethylene compositions

L30 ANSWER 6 OF 6 CA COPYRIGHT 2002 ACS

TI Selective hydroxyl reactivity in methyl .alpha.-D-glucopyranoside

=>

37 11 (EXTRUD? OR EXTRUSION) (P) (ENZYME# OR PROTEASE# OR AMYLASE# OR
LIPASE# OR CELLULASE# OR GLYCOL ETHER# OR BUTOXY? OR
GLYCOL#) (P)
 (ANIONIC OR NONIONIC OR ETHOXY? OR ALKYL BENZENE OR ABS OR HLAS
OR AES OR SULFONATE# OR SULPHONATE#) (P) (HYDRATE# OR DIHYDRATE#
OR TRIHYDRATE# OR ?HYDRATE)

=> d 1-11 137 ti

L37 ANSWER 1 OF 11 USPATFULL

TI Stability enhancing formulation components, compositions and laundry
 methods employing same

L37 ANSWER 2 OF 11 USPATFULL

TI Detergent tablet containing bleach activator of specific particle size

L37 ANSWER 3 OF 11 USPATFULL

TI Detergent compositions containing enduring perfume

L37 ANSWER 4 OF 11 USPATFULL

TI Amorphous alkali metal silicate compound

L37 ANSWER 5 OF 11 USPATFULL

TI O-substituted N,N-diacylhydroxylamine bleach activators and
compositions
 employing the same

L37 ANSWER 6 OF 11 USPATFULL

TI Detergent composition containing optimally sized bleach activator
particles

L37 ANSWER 7 OF 11 USPATFULL

TI Detergent composition containing cylindrically-shaped bleach activator
extrudates

L37 ANSWER 8 OF 11 USPATFULL

TI Process for the manufacture of surfactant cleansing blocks and
compositions thereof

L37 ANSWER 9 OF 11 USPATFULL

TI Conductive N-alkyl polyamide having units containing phosphonium
sulfonate groups

L37 ANSWER 10 OF 11 USPATFULL

TI Conductive aliphatic polyester or polyetherester having units
containing
 phosphonium sulfonate groups

L37 ANSWER 11 OF 11 USPATFULL

TI Liquid or pasty dentifrice and process for its preparation

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